## Practical Application of Quality by Design Principles to The Development of an HPLC Assay Method for an API and Impurities

Kate E. Monks,<sup>1</sup> Imre Molnár,<sup>1</sup> Hans-Jürgen Rieger<sup>1</sup> and François Vogel<sup>2</sup> <sup>1</sup>Molnár-Institute for Applied Chromatography, Berlin, Germany, <sup>2</sup>Novartis Pharma AG, Basel, Switzerland.

A method for the assay of an active pharmaceutical ingredient (API) and five known and unknown impurities was developed in accordance with Quality by Design (QbD) principles. Highly influential separation parameters were simultaneously and systematically studied so that the quality of the HPLC method could be understood, controlled and ensured. Consequently, during routine analysis out of specification (OOS) results can be corrected more effectively.

#### Introduction

The application of Quality by Design and "do it right first time" principles to the development of HPLC methods is notably on the increase, particularly within the field of pharmaceutical analysis.<sup>1–5</sup> A systematic and science-based approach, beginning with predefined goals and focusing on understanding<sup>6</sup> ensures higher quality, more robust and reliable HPLC methods that leads to fewer analysis failures and better method transfer.<sup>7–9</sup>

A key step in guaranteeing the quality of an HPLC method is to investigate the influence of critical parameters on the selectivity of the separation. A tool commonly employed in such systematic studies is chromatography modelling software that maximizes the information content of a limited amount of experimental data.<sup>10</sup> Optimum run conditions and their tolerance to change can be predicted, visualized and examined within resolution models, considerably reducing the necessary investment in developing a robust set of separation conditions.

News

CHROMacademy

Monks



Contents



**Meeting Preview** 

tal Zoo/Getty Imag

Market Trends & Analysis

#### **Experimental**

Eluents Eluent A: phosphate buffer 10 mM with different pH values Eluent B: acetonitrile (AN), methanol (MeOH) and ternary mixtures Flow-rate: 0.8 mL/min Column: YMC (stagroma Equipment AG, Reinach, Switzerland) ODS-AQ (150 mm  $\times$  $3 \text{ mm}, 3 \mu \text{m}$ Instrument: Agilent 1200 Quaternary, dwell volume: 1 mL, extracolumn volume: 0.017 mL, injection volume:  $5 \mu L$ Detection: 243 nm Sample System suitability test mixture solution with four qualified by-products and one unknown, sample concentration: 0.5 mg/5 mL Software Chromatography modelling software: DryLab 4 v.1 (Molnár-Institute, Berlin, Germany) Chromatography data system: Chromeleon 6.8 (Thermo Fisher, Germering, Germany)

#### Experiments for modelling

20 min, t<sub>G2</sub>: 60 min (10→80 %B) Temperature (T): T<sub>1</sub>: 30 °C, T<sub>2</sub>: 50 °C pH of eluent A (pH):  $pH_1$ : 2.2, pH<sub>2</sub>: 3.0, pH<sub>3</sub>: 3.8 Ternary eluent composition (t<sub>c</sub>):  $t_{c1}$ : AN, t<sub>C2</sub>: AN:MeOH (1:1, V:V), tc3: MeOH Cube A: t<sub>G</sub>-T-t<sub>C</sub> at pH 3.0 (12 experimental runs) Cube B: t<sub>G</sub>-T-pH in AN (12 experimental runs)

Gradient time (t<sub>G</sub>): t<sub>G1</sub>:

#### **Results and Discussion**

The first step in this work was to clearly define method goals; next candidate method conditions were systematically investigated with the aid of chromatography modelling software. Resolution/ design spaces and robustness spaces were graphically depicted and final method conditions with tolerance limits were ascertained. Finally, a comprehensive multifactorial robustness evaluation was performed initially in silico and then verified experimentally.

**Method intent**: The aim of this study was the development of a quality, robust HPLC method for the assay of an API and five

Monks

# MUCH MORE THAN YOU WOULD EVER EXPECT!

### **Field-Flow Fractionation**

- → High Resolution Separation of Proteins, Polymers, Particles
- ➤ Ultra-Wide Separation Range 10<sup>3</sup> 10<sup>9</sup> Da, 1nm 100µm
- → Gentle Size Fractionation without Stationary Phase
- Molar Mass, Particle Size by Light Scattering

Polymer Separator

News

Postnova is the world's leading FFF manufacturer with over 25 years experience. Our innovative, patented technology was invented in the USA, is engineered in Germany and is provided to researchers worldwide. Postnova is proud of its excellent customer support.



**CHROMacademy** 



CF2000 Centri FFF Particle Separator



Contents











North America: Tel. +1 (801) 521 2004 Central EU: Tel. +49 (8191) 428 181 Northern EU: Tel. +358 (9) 85 45 510

AF2000 Flow FFF Protein Separator

info@postnova.com www.postnova.com

## Market Trends & Analysis

impurities. Baseline separation or better (Rs.crit  $\geq$  1.5) for all peaks, within a minimum analysis time was required.

Method design and selection: The four critical separation parameters gradient time  $(t_G)$ , temperature (T), pH of eluent A (pH) and ternary eluent composition  $(t_c)$  were selected for systematic evaluation. First the influence on relative retention of  $t_{G}$ , T and  $t_{C}$  was investigated in a simultaneous fashion at a fixed value of pH (pH 3.0) by means of a 3D resolution map (Cube A). Once an appropriate ternary eluent composition was determined (acetonitrile), a further 3D resolution map modelling t<sub>G</sub>, T and pH was constructed (Cube B). Parameter settings and ranges were selected in accordance with the software's guidelines and the generic design of experiments (DoE) is shown in Figure 1.

Resolution models map the critical resolution for each combination of the study parameters (i.e., t<sub>G</sub>, T, pH, t<sub>C</sub>). The value of the critical resolution (R<sub>s.crit</sub>) is represented in colour so that warm colours show large R<sub>s.crit</sub> values and cold colours show low values corresponding to inefficient separations. Specifically, in red regions the resolution is baseline or above ( $R_{s,crit} \ge 1.5$ ) and dark blue lines signalize peak overlaps (Rs.crit

0). Models are calculated from retention data and experimental conditions and can be fine tuned further by entering peak width data and/ or tailing factors, though this was not done in this study.

Contents

The resulting resolution models for Cube A and Cube B are shown in Figure 2 (a) and (b) respectively.

By removing all working points (i.e., combinations of measured parameters) with a critical resolution below the threshold of 1.5 ( $R_{s,crit}$  < 1.5) from the resolution maps, robustness regions were identified and the robustness of the separation was visualized as red irregular geometric bodies. The robustness space for Cube A and Cube B are shown in Videos 1 and 2

#### **Click here for Video 1.**

Video 1: Dynamic view of Cube A robustness

Figure 1: DoE for 3D resolution map construction. Т

t<sub>G</sub>

## Kromasil HILIC

 Ideal for polar compounds MS compatible





1 lot

#### Monks

Last call for the introduction offer on www.kromasil.com/hilic

Spherical silica for your HPLC

Market Trends & Analysis

space. Red regions represent robust above baseline separations.

#### Click here for Video 2.

Video 2: Dynamic view of Cube B robustness space. Red regions represent robust above baseline separations.

A number of different potential working points fulfilling the method goals can be observed within the robustness spaces. From these, the final working point  $t_G$ : 15 min, T: 33 °C, t<sub>C</sub>: AN, pH 3.0 was selected due to the reasonable size of its surrounding robust region and its relatively short run time. A comparison between the predicted chromatogram and experimental verification under these conditions is given in Table 1.

Method evaluation: The selected working point was next subjected to a further multifactorial robustness evaluation. First, the tolerance to changes in additional experimental parameters was investigated *in silico* by studying the influence of six factors simultaneously, namely t<sub>G</sub>, T, pH, flow-rate, start %B and end %B. This was

Figure 2: 3D resolution maps for (a) cube A modelling t<sub>c</sub>, T, t<sub>c</sub>, R<sub>s cut</sub> at constant pH (pH 3.0) and (b) cube B modelling t<sub>c</sub>, T, pH, R<sub>s cit</sub> at constant t<sub>c</sub> (t AN).





## Anticipating reality DryLab meets Nexera UHPLC

...and method development is easier than ever before. Shimadzu and the Molnár Institute cooperate in applying the DryLab®2010\* software and the new generation of the Shimadzu Nexera series. DryLab simplifies and speeds up the process of developing excellent chromatographic separations. Users can model changes in conditions on their PC - and anticipate reality.

- Simulation of more than 1,000,000 separation profiles based on twelve experiments
- Precise prediction of the real separation proven by a large number of users

- Safety and reliability authorities
- Molnár Institute

www.shimadzu.eu



#### Monks



## Market Trends & Analysis

accomplished without the need for additional experimentation by using simulated data from created models to perform full factorial robustness evaluations, registering responses including critical resolution (R<sub>s,crit</sub>) and critical peak pairs. The simulated robustness data was also displayed in a regression coefficient plot (Figure 3), showing that, for the studied

#### Figure 3: From left to right: robustness evaluation and regression coefficients plot.



Table 1. Companson of prediction and experimental retention times for the final working be	Table	e 1:	Comparison of	prediction and	experimental	l retention t	times fo	or the final	working poi
--------------------------------------------------------------------------------------------	-------	------	---------------	----------------	--------------	---------------	----------	--------------	-------------

	F			
Peak Name	Prediction	Experimental	Difference <sup>a</sup>	% Error <sup>b</sup>
Unknown	8.18	8.25	0.07	0.86
501	9.44	9.46	0.02	0.21
Main (API)	9.78	9.80	0.02	0.20
502	10.08	10.08	0.00	0.00
512	10.52	10.52	0.00	0.00
374	10.95	11.05	0.10	0.91

<sup>a</sup> Difference = Experimental–Prediction

<sup>b</sup> % Error = [(Experimental–Prediction)/Prediction]×100

Contents







News



## Market Trends & Analysis

tolerance windows, the factor most strongly influencing resolution is the pH of the eluent A.

Once the working point and tolerances were approved within the software, an experimental study summarized in Table 2 was effectuated to confirm results.

The final experimental parameter settings and tolerances for this assay were  $t_G$ : 15 1 min, T: 33 2 °C,  $t_C$ : AN and pH: 3.1 0.1.

With the aid of the software we can visually observe that for the pH another value other than 3.0 would be better selected: The value of 3.1. At this pH both neighbouring peaks of the API (MAIN) are moving away from the API peak. If the pH value is 2.9, both peaks are moving towards the API peak. Furthermore, we can observe

Table 2: Experimental robustness evaluation.(a) Parameter settings and levels used in the experimental robustness study.

		-	0	+
Α	t <sub>G</sub> (min)	14	15	16
В	T (°C)	31	33	35
С	рН	2.9	3.0	3.1

<b>2(b)</b> DoE design table (randomized).												
	1	2	3	4	5	6	7	8	9	10	11	12
Α	0	0	-	+	0	-	0	+	+	-	-	+
В	0	0	+	-	0	-	0	-	+	-	+	+
С	0	0	+	-	0	+	0	+	-	-	-	+

in a visual way in DryLab that other factors do not influence the selectivity considerably. This opens up new possibilities to deal with out of specification results more flexibly, as we can clearly see what has to be done to correct an OOS case. With regards to judging robustness issues, we have to look at the peak movements and understand which factors provoke which changes in critical peak positions, i.e., in critical resolution.

#### Summary

Quality by Design and "do it right first time" principles were successfully employed to develop a robust HPLC assay method in a transparent and understandable way with the assistance of chromatography modelling tools and an appropriate workflow.

#### References

- M. Schweitzer, M. Pohl, M. Hanna-Brown, P. Nethercote, P. Borman, G. Hansen, K. Smith and J. Larew, *Pharmaceutical Technology Europe*, **22**(2) 29–36 (2010).
- F.G. Vogt and A.S. Kord, J. Pharm. Sci., 100(2) 797–812 (2011).
- P. Borman, J. Roberts, C. Jones, M. Hanna-Brown, R. Szucs and S. Bale, *Separation Science*, 2(7) 2–8 (2010).
- M. Hanna-Brown, P. Borman, S. Bale, R. Szucs, J. Roberts and C. Jones, *Separation Science*, 2(1) 12–20 (2010).
- K.E. Monks, I. Molnár, H.-J.Rieger, B. Bogáti and E. Szabó, *J. Chromatogr. A*, **1232**, 218–230 (2012).



## Market Trends & Analysis

	ulto			
Exp.	t, (min)	T (°C)	рH	Experimental chromatogram
1	15	33	3.0	12 12 Main 12 12
2	15	33	3.0	21 0 1 8 200 4 200 4 200 4 200 4 200 4 200 4 200 4 200 4 200 4 200 4 200 4 200 4 200 4 200 4 200 4 200 4 200 4 200 4 200 4 200 4 200 4 200 4 200 4 200 4 200 4 200 4 200 4 200 4 200 4 200 4 200 4 200 4 200 4 200 4 200 4 200 4 200 4 200 4 200 4 200 4 200 4 200 4 200 4 200 4 200 4 200 4 200 4 200 4 200 4 200 4 200 4 200 4 200 4 200 4 200 4 200 4 200 4 200 4 200 4 200 4 200 4 200 4 200 4 200 4 200 4 200 4 200 4 200 4 200 4 200 4 200 4 200 4 200 4 200 4 200 4 200 4 200 4 200 4 200 4 200 4 200 4 200 4 200 4 200 4 200 4 200 4 200 4 200 4 200 4 200 4 200 4 200 4 200 4 200 4 200 4 200 4 200 4 200 4 200 4 200 4 200 4 200 4 200 4 200 4 200 4 200 4 200 4 200 4 200 4 200 4 200 4 200 4 200 4 200 4 200 4 200 4 200 4 200 4 200 4 200 4 200 4 200 4 200 4 200 4 200 4 200 4 200 4 200 4 200 4 200 4 200 4 200 4 200 4 200 4 200 4 200 4 200 4 200 4 200 4 200 4 200 4 200 4 200 4 200 4 200 4 200 4 200 4 200 4 200 4 200 4 200 4 200 4 200 4 200 4 200 4 200 4 200 4 200 4 200 4 200 4 200 4 200 4 200 4 200 4 200 4 200 4 200 4 200 4 200 4 200 4 200 4 200 4 200 4 200 4 200 4 200 4 200 4 200 4 200 4 200 4 200 4 200 4 200 4 200 4 200 4 200 4 200 4 200 4 200 4 200 4 200 4 200 4 200 4 200 4 200 4 200 4 200 4 200 4 200 4 200 4 200 4 200 4 200 4 200 4 200 4 200 4 200 4 200 4 200 4 200 4 200 4 200 4 200 4 200 4 200 4 200 4 200 4 200 4 200 4 200 4 200 4 200 4 200 4 200 4 200 4 200 4 200 4 200 4 200 4 200 4 200 4 200 4 200 4 200 4 200 4 200 4 200 4 200 4 200 4 200 4 200 4 200 4 200 4 200 4 200 4 200 4 200 4 200 4 200 4 200 4 200 4 200 4 200 4 200 4 200 4 200 4 200 4 200 4 200 4 200 4 200 4 200 4 200 4 200 4 200 4 200 4 200 4 200 4 200 4 200 4 200 4 200 4 200 4 200 4 200 4 200 4 200 4 200 4 200 4 200 4 200 4 200 4 200 4 200 4 200 4 200 4 200 4 200 4 200 4 200 4 200 4 200 4 200 4 200 4 200 4 200 4 200 4 200 4 200 4 200 4 200 4 200 4 200 4 200 4 200 4 200 4 200 4 200 4 200 4 200 4 200 4 200 4 200 4 200 4 200 4 200 4 200 4 200 4 200 4 200 4 200 4 200 4 200 4 200 4 200 4 200 4 200 4 200 4 200 4 200 4 200 4 200 4 200 4 200 4 200 4 200 4 200 4 200 4 200 4 200
3	14	35	3.1	201 8 501 8 502 Main 512 74 512 12 512 12
4	16	31	2.9	12 microsoft and a second and a
5	15	33	3.0	212 Main 502 Main 512 Main
6	14	31	3.1	10 12 12 12 12 12 12 12 12 12

10

24

News

CHROMacademy







Contents



**Meeting Preview** 

## Market Trends & Analysis

## Training & Events

25

2(c) Results. (Continued)								
Exp.	t <sub>G</sub> (min)	T (°C)	рН	Experimental chromatogram				
7	15	33	3.0	8 bunknown 501 512 512 512 512 514				
8	16	31	3.1	∞ -   >unknown 501 - 501 - 512 512 Main				
9	16	35	2.9	8 9 1 1 1 1 1 1 1 1 1 1 1 1 1				
10	14	31	2.9	8 - Hunknown 6				
11	14	35	2.9	8 - bunknown 6 - bunknown 502 Main 512 Main 512 Main				
12	16	35	3.1	202 Main 502 Main 512 Main				

- 6. ICH Q8 (R2) Guidance for Industry, Pharmaceutical Development (2009).
- 7. I. Molnar, H.-J Rieger and K.E. Monks, J. Chromatogr. A, 1217(19), 3193 (2010).
- 8. K.E. Monks, H.-J. Rieger and I. Molnár, J. of Pharma. Biomed. Anal., 56(5), 874 (2011).
- 9. I. Molnár, K.E. Monks, H.-J. Rieger and B.-T. Erxleben, The Column, 7(5), 2 (2011).
- 10. L.R. Snyder, J.J. Kirkland and J.L. Glajch, Practical HPLC Method Development, 2nd ed., Wiley-Interscience, New York, USA (1997).

Kate E. Monks received her degree in chemistry in Valencia, Spain. She worked at the Faculty of Pharmacy at the Charles University in Prague in the area of synthesizing drug candidates for tuberculosis. Since 2008 she has been working at the Molnár Institute in Berlin, specializing in DryLab 3D method development.

Imre Molnár received his PhD from the University of Saarbrücken, Germany, with special training in analytical chemistry. He then worked for two years as a postdoctoral fellow with Csaba Horváth at Yale University, USA, on fundamentals of reversed-phase chromatography. He founded the Molnár Institute 30 years ago and developed the DryLab software with L.R. Snyder, J. Dolan, T. Jupille and their team. He has considerable

E-mail:

kate.monks@molnar-institute.com www.molnar-institute.com



experience in the development of HPLC instruments, column production and HPLC applications and is specialized in pharmaceutical and biopolymer research and analysis.

Hans-Jürgen Rieger received his PhD from the Free University of Berlin, Germany. He started to work for the Molnár Institute in 1999 as an application chemist with a specialization in software programming and currently holds the position of software product manager. In cooperation with Dr Molnar, he is responsible for the development of new versions of DryLab and the software tool PeakMatch.

François Vogel holds a swiss federal degree as technician as well as an economic degree and project management diploma (IPMA C). He has over 15 years of experience in the pharmaceutical industry. He has made significant contributions in quality by design, reversed phase, HILIC and method development in the Novartis Analytical Network and is a champion of driving the application of DryLab optimization software throughout all pharmaceutical teams in Basel. Since 2006, he has been a trainer in analytics, HPLC and project management In his day-to-day work, he specializes in HPLC method development, trouble-shooting and structure elucidation by LC-MS.